



# GENESIS AND STABILITY OF TRIBOLAYERS IN SOLID LUBRICATION: CASE OF PAIR DLC-STAINLESS STEEL\*

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## Abstract

The morphology, dimensions and chemical composition of tribolayers strongly depend on the pressures and temperatures acting on the contact. They are formed by reactions between the surfaces in contact with each other as well as with the atmosphere, lubricants and possible contaminants. In this paper the influence of test time (180, 500, 1000 and 2500 h) into the formation and characteristics of tribolayers in pairs DLC-stainless steel tested under refrigerant gas R134a atmosphere without presence of lubricating oil was analyzed. The characterization was performed using scanning electron microscopy, energy dispersive spectroscopy (SEM-EDS) (morphology and chemical composition) and white light interferometry (dimensions). The tribolayers thicknesses ranged from 100 to 500 nm and they were composed by elements originated from mutual transfers between the tribological pairs as well as oxides, being more pronounced on the stainless steel surface. The results show that the tribolayers are chemically stable (maintained the same composition over time) and the thickness remained stable after a thousand hours of testing.

**Keywords:** Tribology; Tribolayer; Diamond-like carbon (DLC).

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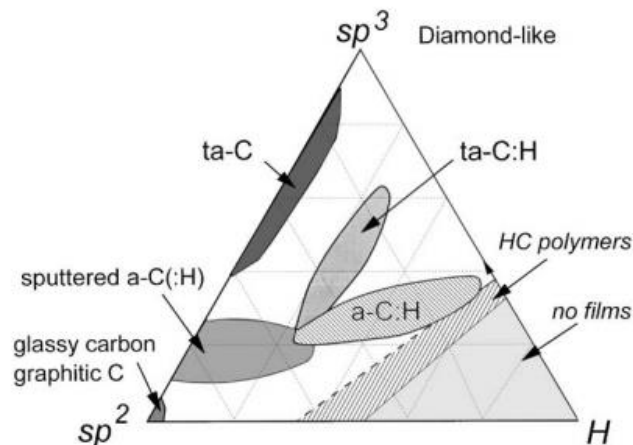
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## 1 INTRODUCTION

Considerable fraction of the energy generated in the world is lost due to friction and wear in mechanical systems. According to Jost [1] it is possible to reduce these losses by up to 20% only applying existing knowledge and techniques.

Fluid lubrication is widely applied in order to reduce friction and wear, however, environmental factors suggest the industry to reduce or even eliminate the lubricating oils. In addition, there are applications where the oil presence is not possible, for example, environments under high temperature and pressure which can promote chemical degradation of the oil. Contaminant-free systems such as those used for food and pharmaceuticals equipment are also instances where dry lubrication is potentially interesting [2]. Therefore, improving the tribological performance of mechanical elements is a prerequisite for new products development and manufacturing processes [3].

Oil less tribo systems represent a new challenges for the science and surface engineering, stimulating new projects on coatings and self-lubricant material. Amorphous carbon coatings, known as diamond-like carbon (DLC) have been an alternative in many technological applications, such as hard coating for tools, automotive parts, computer hard drives and micro electro mechanical systems (MEMS). The applications are mainly associated with its high hardness, high chemical inertness and solid lubrication capacity [4]. The DLC family consist of amorphous carbon hydrogenated alloys (a-C:H), as shown in the ternary diagram on figure 1, where different families of DLC are characterized by the percentage of hybridizations  $sp^2$ ,  $sp^3$  and amount of hydrogen [5]. In addition, doping elements are incorporated into DLC to modify its properties, such as silicon (a-C:H:Si) which decreases the free energy and the residual stress of the coating modifying its tribological behavior [6]. Other common dopant elements in DLC are tungsten, titanium, molybdenum and chromium.



**Figure 1:** Ternary phase diagram for hydrogenated amorphous carbon alloys [3].

The low coefficient of friction associated with DLC is often associated with tribolayer formation, which is a result mainly of transfer material from DLC to the counter body and vice versa [6-8]. The genesis of these tribo layers is strongly dependent on environment. For atmospheres such as air, oxygen or nitrogen, the increase of humidity usually reduces coefficient friction of DLC-steel contact [9,10]. In a study performed by de Mello, Binder et al. [11], it was shown that the presence of protective atmosphere (R600a) modifies the chemical structure of tribolayer generated on the counter body, reducing the friction coefficient and the wear rate of

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the system (DLC / steel AISI 52100). In another work performed by Demas and Polycarpou [12] pin-on-disc tests were performed with gray cast iron (pin and disc) without lubrication and different atmospheres (O<sub>2</sub>, CO<sub>2</sub>, N<sub>2</sub>, Ar and R134a). Under O<sub>2</sub> atmosphere, the wear is ruled by the oxidation that occurs on the wear track, whereas in air and R134a it also contains an adhesive component. Under N<sub>2</sub> atmosphere, the wear is dominated by adhesion, while in CO<sub>2</sub> there is a slight polishing on the surface, removing only superficial asperities, thus leading to the best tribological performance among the tested atmospheres.

The tribo layers settle on the real contact area, therefore, rule the tribological behavior of tribo systems. Their dimensions are reduced (nanometers), and this factor hinders to understand their formation and stabilization. Tribo layers are formed from physicochemical interactions between the surfaces in contact and relative motion, which ranges from mutual transfer of materials and reactions between atmosphere, lubricants and contaminants present in the contact [13].

The contact conditions affect the formation and destruction of tribo layers, i.e., the stability of tribo layers depends on contact evolution. When the formation and destruction rates of tribo layers acquire the same values, the tribo layers achieve a certain mechanical stability, which from the point of view of tribological performance is fundamental to stability contact.

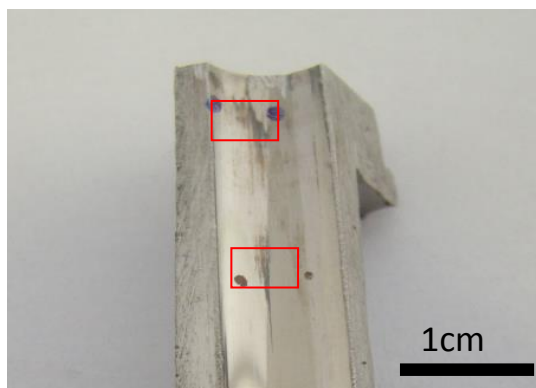
This study aims to understand the genesis of tribo layers formed in DLC-stainless steel pair with reciprocating movement and controlled atmosphere, as well as to evaluate their mechanical and chemical stability.

## 2 MATERIALS AND METHODS

The tested tribological pairs was formed by a piston and cylinder coaxially coupled, where the cylinder is made of stainless steel 304 and the piston of AISI 52100 steel coated with chromium nitride (CrN) (thickness of 1.5-2.0 μm) and DLC (thickness ~ 1.0 μm). The experiments imposed a reciprocating movement (1.7 mm amplitude and frequency of 350 Hz) in a tetrafluoroethane (R134a) gas atmosphere.

Nine experiments were conducted, being three of them with 180 hours duration and two more for each period of 500, 1000 and 2500 hours.

After tests, the stainless steel cylinders were sectioned in order to access the contact areas. The body and counter body surfaces were cleaned with cotton and absolute ethyl alcohol. At a first glance, darker regions were observed (highlighted in figure 2) where, in principle, it would be easier to find tribo layers.

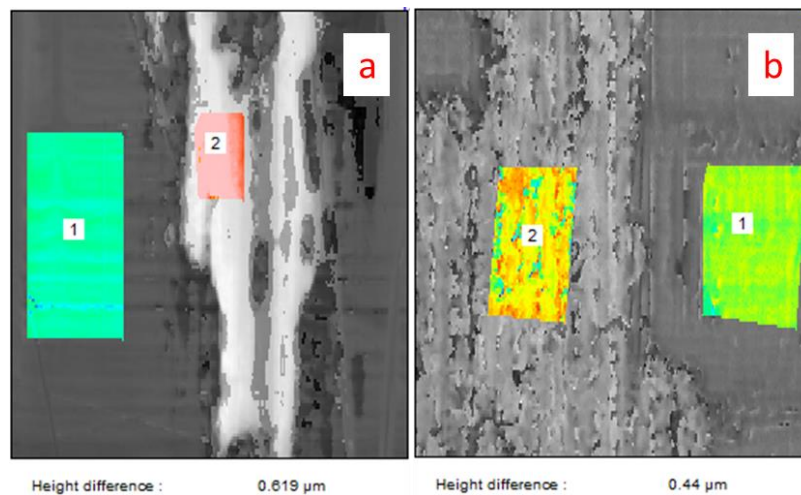


**Figure 2:** Possible tribo layers highlighted in red.

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The relevant regions were analyzed by using Scanning Electron microscopy (SEM-JEOL JSM-6390LV).

For data acquisition of dimensional analysis, the white light interferometer (Zygo New View 7300) was used. MountainsMap<sup>®</sup> software was used to perform the measurement of the average tribo layers thicknesses. The procedure consists in selecting two regions, one within the tribolayer and another on the original surface, and then the average height difference between these two regions can be calculated. On figure 3 there are two examples of thickness measurement of tribo layers for two different periods tests (500 and 2500 hours). Regions labeled 1 refer to the original surface and label 2 refer to the tribolayer. The measurements were repeated five times at different points.



**Figure 3:** Examples of tribo layers thickness measurement on the stainless steel surface. (a) 500 hours (b) 2500 hours.

### 3 RESULTS AND DISCUSSION

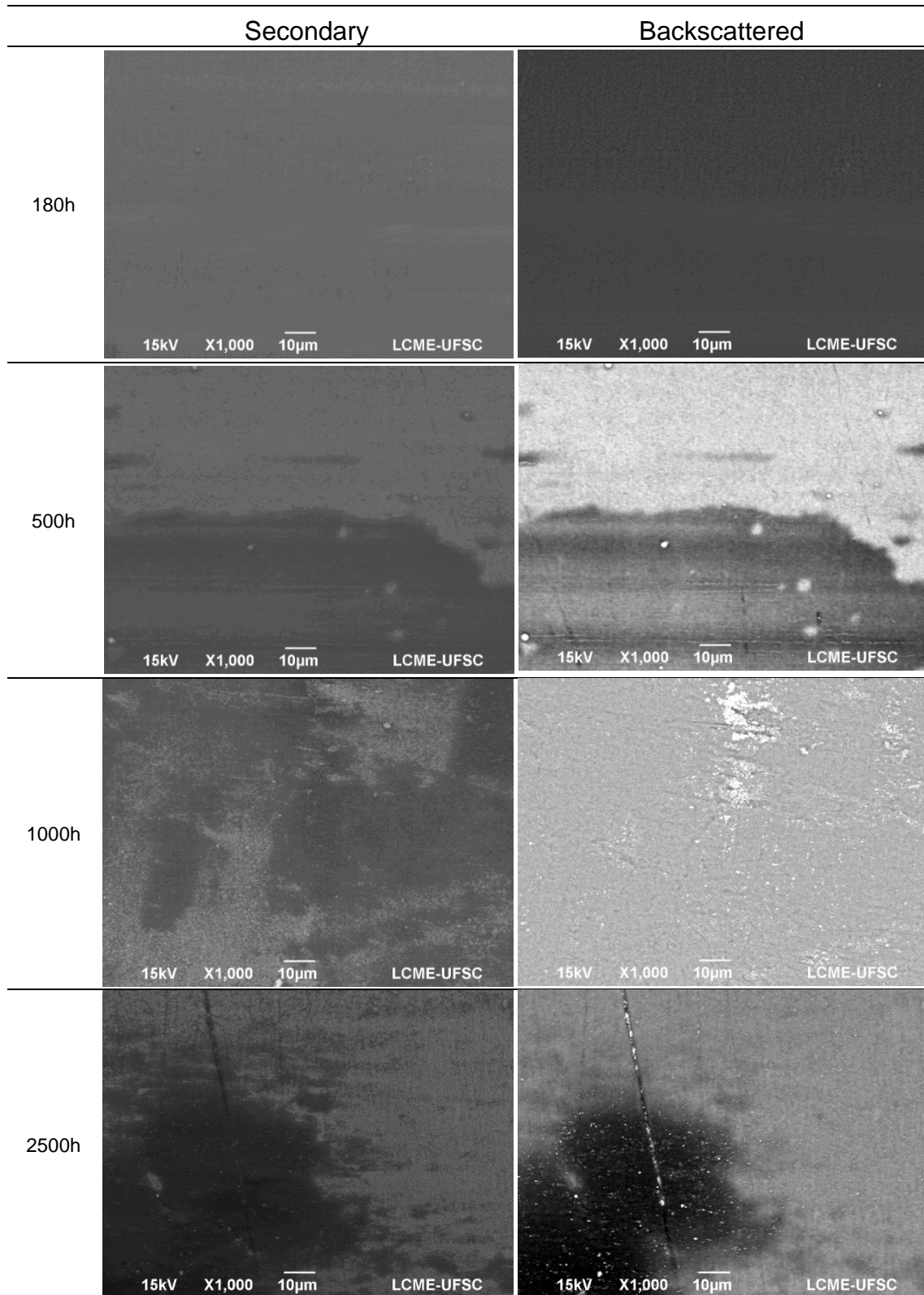
#### 3.1 SEM and EDS

Figures 4 and 5 show typical tribolayers present in DLC and stainless steel surfaces, respectively. In the right columns, the images were obtained by using backscattered electron and evidence variations in average atomic number.

Although tribolayers are found in both body and counterbody, they were more pronounced on the stainless steel surfaces.

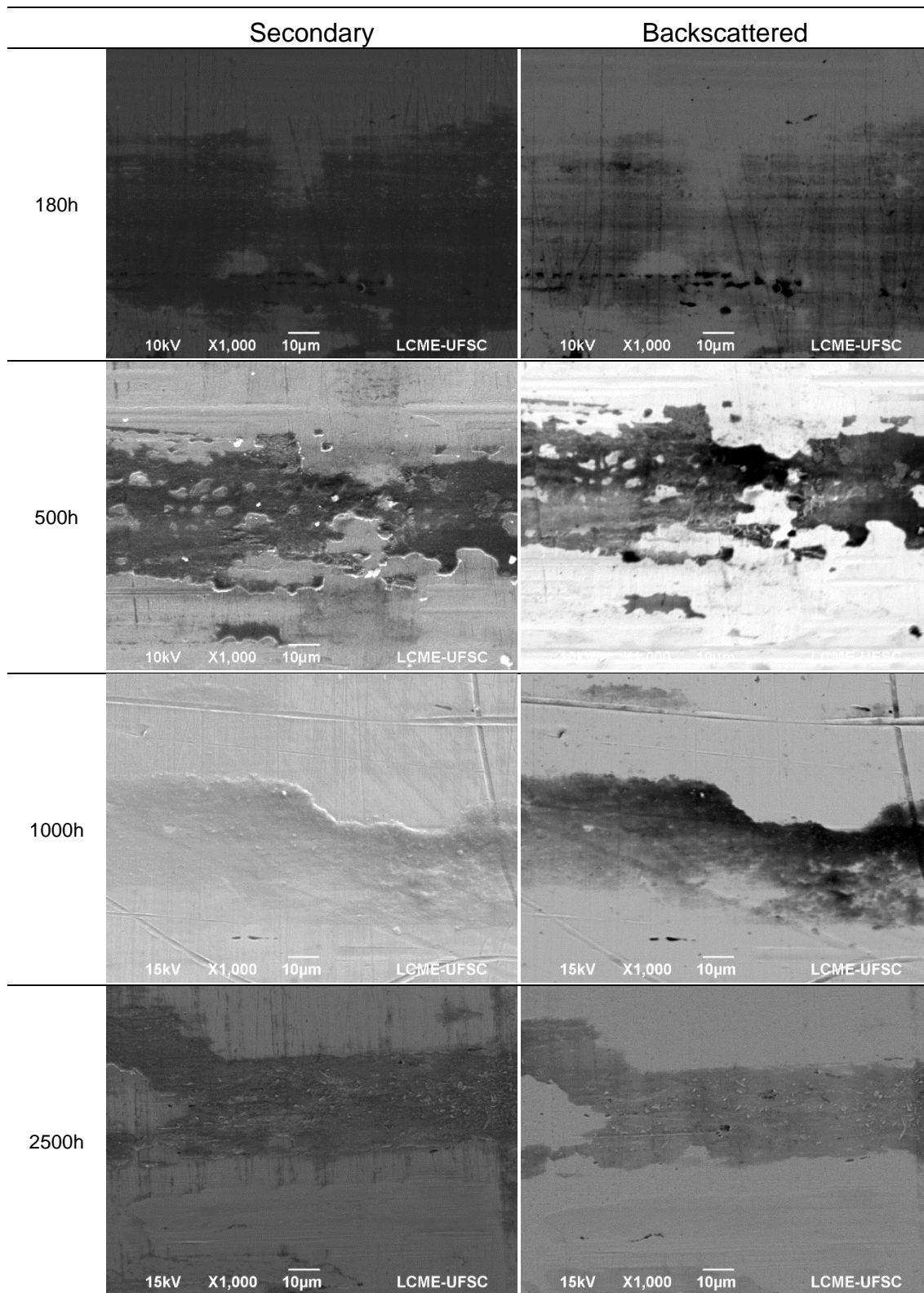
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**Figure 4: Scanning electron microscopy (SEM) images of tribolayers found in DLC.**

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**Figure 5:** Scanning electron microscopy (SEM) images of tribolayers found in stainless steel.

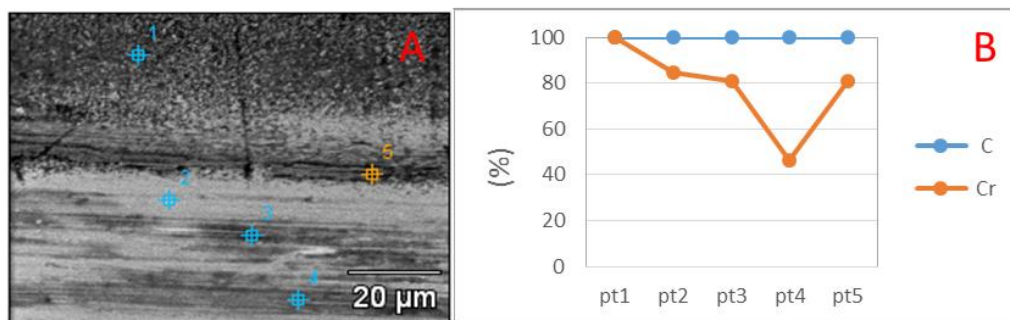
The variation of concentration of carbon, chromium and oxygen outside and inside the tribolayer is displayed in figures 6 and 7. These values were obtained dividing the measured values by the highest concentration for each element (normalization). The carbon percentage on DLC surface varied very little (Figure 6) whereas carbon variation in the tribolayer associated with the stainless steel surfaces is very

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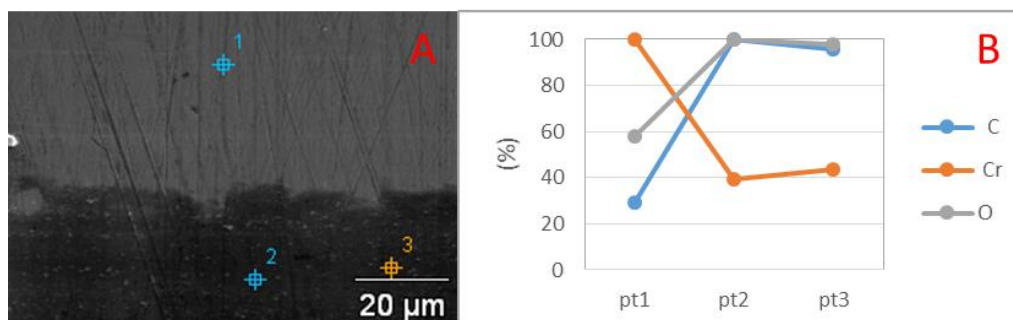


important, figure 7, and its origin may be associated to material transfer from DLC coating. It is also noticeable the high oxygen of the tribolayer clearly indicating an oxidation process in accordance with Wu, Pai and Hon results [14].

Due to electron beam penetration, the substrate has a direct influence on EDS results. Thereby, reduction of the chromium concentration on the rubbed regions of DLC occurred due to coating thinning, which allowed more interaction between the electron beam and the substrate atoms (AISI 52100) below the CrN layer. The chromium concentration on tribolayers associated with the stainless steel decreased when compared to the original surface (Figure 7).



**Figure 6:** Variation of the carbon and chromium concentration outside (1) and inside (2, 3, 4, 5) of tribolayer over DLC. (A) Measurements location. (B) Evolution of C and Cr concentration.

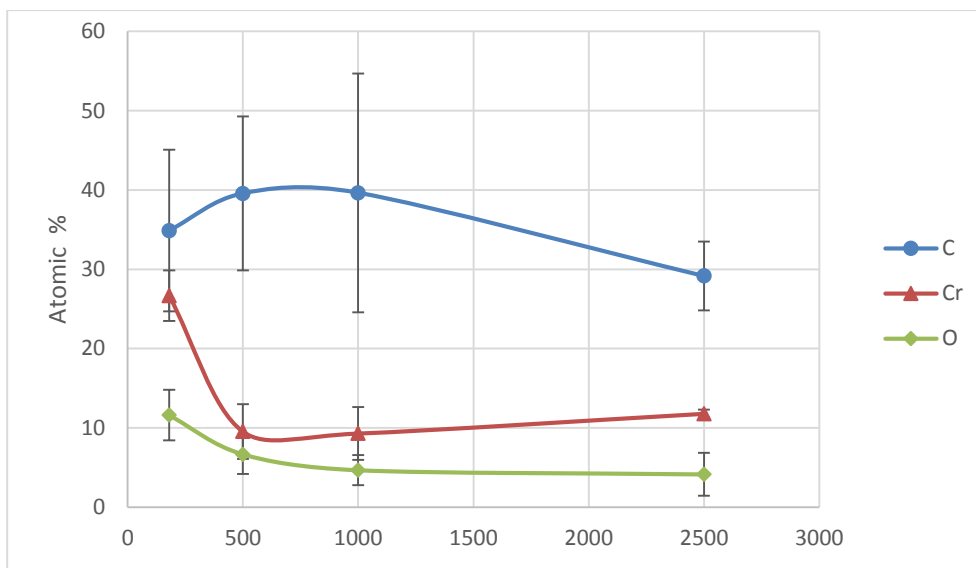


**Figure 7:** Variation of the carbon, chromium and oxygen concentration outside (1) and inside (2, 3) of tribolayer over stainless steel. (A) Measurements location. (B) Evolution of C, Cr and O concentration.

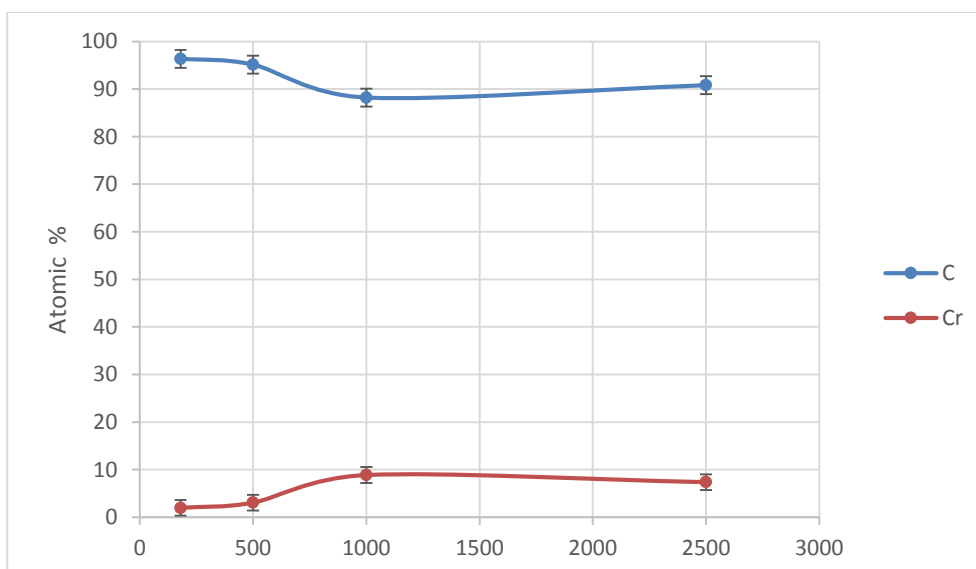
Figure 8 shows the evolution of carbon, chromium and oxygen concentrations with test time for tribolayers related to stainless steel surfaces. The atomic percentage of oxygen slightly decreased for test times up to 1000 hours and after that the values remain almost constant. The same behavior is observed for chromium, however, the initial drop is more abrupt and occurs between 180 and 500 hours. In spite of the large dispersion presented by the carbon amount, it is possible to observe a reverse trend in relation to chromium evolution, indicating the formation of carbon-rich tribolayers originated from the DLC coating. The figure 9 shows the evolution of carbon end chromium concentrations with test time for tribolayers related to DLC surfaces. The same inverse behavior between the carbon and chromium concentration was observed again, indicating a possible mutual transfer of material between the surfaces [15].

In synthesis, tribo-reactions given rise to tribolayers are active up to a thousand hours of testing time. After that time, the tribolayers remain chemically quite stable.

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**Figure 8:** Evolution of carbon, chromium and oxygen concentrations with test time. Tribolayers found on stainless steel surfaces.



**Figure 9:** Effect of test time on carbon and chromium content in tribolayers related to DLC surfaces.

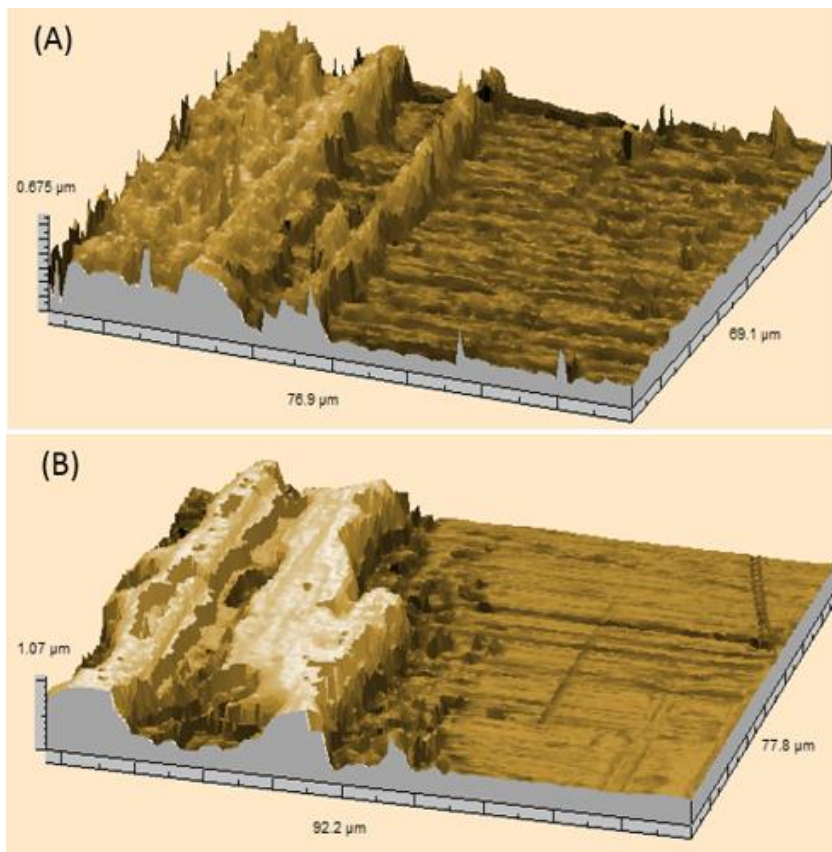
### 3.2 Interferometry

Figure 10 shows the surface topography of tribolayers after 180 and 500 hours of testing. It is noticeable that after 500 hours the tribolayer is thicker and shows large and flat areas, probably due to the relative motion between surfaces in accordance with Olofsson, Gerth et al [16] results.

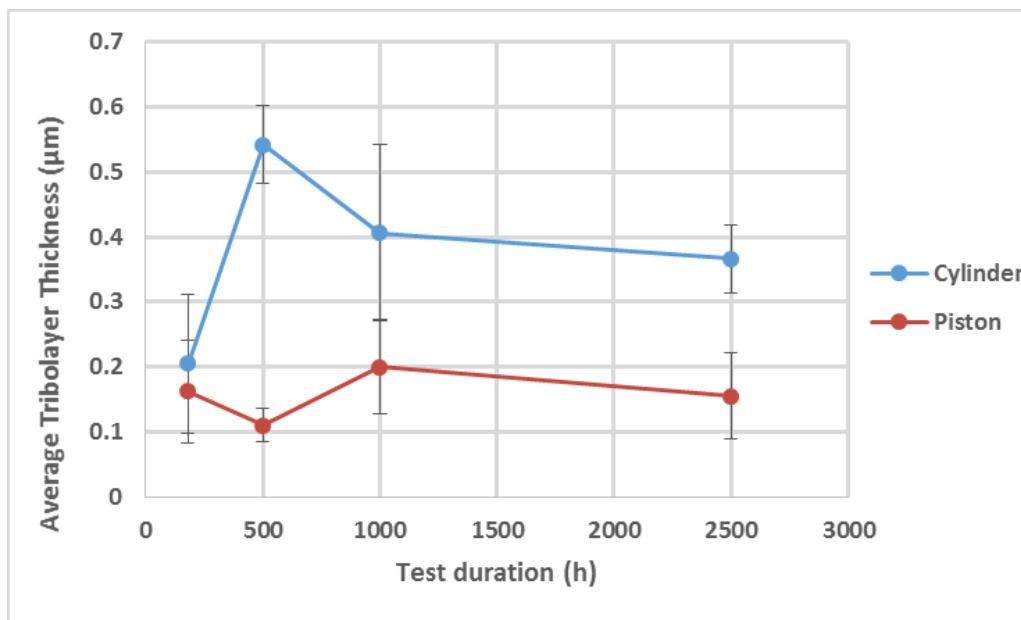
Figure 11 shows the test time influence on tribolayers thicknesses. For long tests times (1000 hours), the average thickness of tribolayers varied little, stabilizing at around 170nm for DLC specimens and 380nm for stainless steel samples. Initially (180 to 500 hours) there is a significant increase in the thicknesses of tribolayers over stainless steel, from 200nm to 500nm, and afterwards (500 to 1000 hours) a decrease to 400nm, whereas the opposite occurs in the DLC. This suggests the formation and destruction of the tribolayer on both the body and the counterbody until 1000 hours, after this time they reach an equilibrium state regarding their thicknesses.

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**Figure 10:** Tribolayers surface topography on stainless steel surface: (A)180 hours (B)500 hours.



**Figure 11:** Tribolayers average thickness on DLC and stainless steel.

#### 4 CONCLUSIONS

The tribolayers were more pronounced on the surfaces of stainless steel. However, The tribolayer composition is associated with the mutual transfer of materials between the surfaces, along with oxidation processes, especially on the stainless steel surface.

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The concentration of chromium, carbon and oxygen presented a significant variation between 180 and 1000 hours. After that time, the tribolayers remain chemically quite stable. In addition, the thickness of tribolayers also varied in the same interval, so that the thicker the layer on the body, the thinner the layer on the counterbody, suggesting a mutual destruction and formation of tribolayers until a stable state is reached.

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